Exploration of Structural Frameworks for Reactive and Enantioselective Peptide Catalysts by Library Screenings

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S Supporting Information

[AB](#page-4-0)STRACT: [By screening](#page-4-0) large-scale N-terminal L-prolyl peptide libraries, we explored efficient catalysts for asymmetric Michael addition of a malonate to an enal. The catalytically active peptides obtained by the screening could be categorized into two groups based on the similarity of amino acid sequences. One group of the peptides selectively gave an Sproduct, whereas the other gave an R-product, despite all of the peptides having a common N-terminal sequence, Pro-D-Pro. Further optimization by second-generation screenings

afforded more reactive and enantioselective catalysts. It was found that the peptides having a histidine residue at the seventh position were good catalysts, and their reaction efficiencies were correlated with the abilities of entrapping a substrate into resin beads.

 \sum eptides are promising candidates for efficient and selective organocatalysts.1,2 One of the distinct features of peptide catalysis is that the structural diversity of peptides can be easily attained by changi[ng](#page-4-0) amino acid sequences. While such a tunability of peptides assures the high potency for the development of capable catalysts, identifying an optimal sequence from numerous combinations of amino acids is difficult and time-consuming. A powerful solution to this problem is the screening of a peptide library in a highthroughput manner. $3,4$ We have recently reported a novel highthroughput screening method, 5 in which resin-bound Nterminal prolyl pept[ide](#page-4-0)s can be assayed for an amine-catalyzed Michael reaction 6 (Scheme 1). The procedure includes the

following steps: (1) stirring a mixture of an enal, a dye-labeled nucleophile, and a peptide library on beads; (2) anchoring the dye-labeled product to the peptide by addition of a reducing agent; (3) collecting colored beads under the microscope; and (4) analyzing sequences by electrospray ionization tandem mass spectrometry (ESI-MS/MS) after the cleavage of peptides from the beads. Because active peptides can be colored through the in situ reductive amination, a premodification of a peptide library is not necessary. By using this method, we successfully obtained a histidine-containing peptide which had the function of enhancing the substrate binding to the catalyst.⁵ In our previous study, however, peptide libraries were constructed based on a reported sequence; only a few residues we[re](#page-5-0) varied, while the main secondary structures such as a β -turn and a helix were retained. It was expected that new structural frameworks for more efficient catalysts could be found by screening peptide libraries with fully randomized residues, which would cause dramatic changes in enantioselectivity due to the diverse threedimensional structures of screened peptides. In this paper, a pair of peptides with complementary enantioselectivity was developed based on the consensus sequences obtained by the screenings and was further optimized with second-generation libraries.

Initially, two kinds of libraries consisting of N-terminal proline and the following variable six residues were constructed by a split-and-mix method $(Figure 1)$. At the C-terminus, octaethylene glycol and methionine were introduced to increase the sensitivity in ESI-MS/MS [analyses](#page-1-0) a[n](#page-5-0)d for the cleavage of peptides from resin beads by cyanogen bromide, respectively. The randomized part of library A contained six kinds of amino

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Figure 1. Peptide libraries with six randomized residues. AA = amino acid.

acids, which were the components of previously developed catalysts.⁵ We expected that a histidine-containing peptide would be obtained, but upon screening this library with 4 nitrocin[na](#page-5-0)maldehyde and dye-labeled malonate 1 (Scheme 2), δ

none of the detected sequences had a histidine residue. Thus, in order to access to a wider conformational space, D-leucine was employed in library B instead of histidine in library A.⁹ The peptide sequences found by the screenings of both libraries are shown in Figure 2, and a number of observatio[ns](#page-5-0) are noteworthy: (i) all of the peptides have D-Pro at the AA^2 position; (ii) D-amino acids appeared in a higher frequency than

	D-Leu	Tyr	Aib	D-Pro	Leu	D-Pro	11
type I	D-Leu	Trp	Aib	D-Pro	Tyr	D-Pro	12
	D-Leu	Trp	Aib	D-Pro	Tyr	D-Pro	13
	D-Leu	Trp	Aib	D-Pro	Trp	D-Pro	14
	Tyr	D-Pro	D-Leu	Tyr	Aib	D-Pro	15
	Trp	D-Pro	D-Leu	Tyr	Aib	D-Pro	16
type II	Leu	D-Pro	D-Leu	Trp	Aib	D-Pro	17
	D-Leu	D-Pro	D-Leu	Leu	Aib	D-Pro	18
	Tyr	D-Pro	D-Leu	Trp	Leu	D-Pro	19
	Tyr	D-Pro	D-Leu	Tyr	Trp	D-Pro	20

Figure 2. Detected sequences by initial screenings. Consensus residues are boxed by blue and red lines.

statistically expected; (iii) all peptides found in library B possess D-Leu; (iv) the majority of the peptides detected from library A have a consensus sequence, Pro-D-Pro-X-D-Pro-X-X-{Aib or D-Pro}; (v) the similar sequences with D -Leu at the AA⁷ position were also found in the peptides from library B; (vi) other peptides obtained from library B have a different consensus sequence, Pro-D-Pro-X-X-D-Leu-D-Pro-X; and (vii) the preserved parts in the consensus sequences from both libraries unexceptionally consisted of D-amino acids or Aib. Consequently, the detected peptides except for samples 9 and 10 could be classified as type I and type II according to the common sequences with D-amino acids (Figure 2).

Four type I peptides (samples 1, 4, 7, and 8) and two type II peptides (samples 15 and 19) were selected, and the corresponding resin-supported catalysts were prepared by a conventional solid-phase peptide synthesis. In the presence of these catalysts, asymmetric Michael addition of dimethyl malonate to 4-nitrocinnamaldehyde was conducted (Table 1).¹⁰ This reaction hardly proceeded with the peptides obtained from colorless beads during the screening (Table 1, entries 1− [3](#page-2-0))[. T](#page-5-0)he reaction was significantly enhanced by the t[ype](#page-2-0) [I](#page-2-0) peptides (Table 1, entries 4−7). The ee val[ues were](#page-2-0) moderate, and the major isomer of the product had the S-configuration. From the [fact that](#page-2-0) no amino acid with the L-form was found at the seventh residue in the type I peptides, D-amino acids seemed suitable at this position. When the seventh Aib of catalyst 2 was replaced with D-amino acids, the reactivity slightly increased (Table 1, entries 8−10), and in the cases with D-Ala and D-Trp, a higher enantioselectivity was also observed (entries 9 and 10). [Next, ty](#page-2-0)pe II peptides were examined. While the reactivity decreased somewhat compared to that of type I peptide 2, the ee values were higher (Table 1, entries 11 and 12). The configuration of the major product was R, which was the opposite enantioselectivity of th[e case w](#page-2-0)ith the type I peptide. This is interesting because the two N-terminal residues of peptides 2 and 3 and even the three residues of peptides 2 and 4 are identical to each other, but the selectivities are completely altered by the subsequent residues.^{3c} The replacement of D-Leu at the fifth residue in peptide 3 to L-Leu decreased the catalytic efficiency along with a dra[mat](#page-4-0)ic loss in the enantioselectivity (Table 1, entry 13). Similar to the type I peptide, the seventh residue of peptide 3 also affected the reaction; the use of ala[nine ins](#page-2-0)tead of tyrosine lowered the conversion (Table 1, entry 14). In our previous study, it was found that a histidyl side chain allocated in proximity to an Nterminal pr[oline en](#page-2-0)hances the reaction.⁵ Accordingly, the seventh tyrosine of peptide 3 was replaced with histidine, which resulted in a marked improvem[en](#page-5-0)t of the reaction efficiency (Table 1, entry 15).

Next, for a further enhancement of the catalytic capability, second-ge[neration](#page-2-0) libraries were constructed based on the initial screenings and evaluations (Figure 3). For the type I peptide, an octapeptide library with four randomized residues was designed (library A-2nd). [The va](#page-2-0)riable positions constituted 10 kinds of amino acids, and the D-forms of amino acids were employed at the seventh residue. A different approach was employed for the type II peptide; additional six residues were introduced to the C-terminus of peptide 5 by using the same components of the initial library (library B-2nd). Such a stepwise introduction of the randomized block facilitates the screening of long peptide sequences by suppressing the scale of a library not to exceed a practically affordable number of beads. These second-generation libraries

^aThe symbol in parentheses indicates the absolute configuration of the major enantiomer. ^bNot determined.

Figure 3. Design of second-generation libraries.

were screened, as well, but the reaction time for coloring was shortened to 10 min from 3 h for the initial screening. Although the detected sequences from library A-2nd did not show a noticeable tendency, the amino acids with bulky side chains such as tyrosine and tryptophan were frequently observed in all of the randomized positions (see Figure S1 in the Supporting Information). The peptides with these sequences significantly accelerated the Michael reaction (Table 2, entrie[s 1 and 2\),](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.6b01591/suppl_file/jo6b01591_si_001.pdf) [compared t](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.6b01591/suppl_file/jo6b01591_si_001.pdf)o peptide 2 (Table 1, entry 4). Among them, peptide 6 containing D-His at the seventh residue was most reactive, which was clearly indicated by the data with 10 mol % of catalyst loading (Table 2, entries 1 and 2, conversions in parentheses).¹¹ The importance of each amino acid of peptide 6 at the AA^6 to AA^8 positions was confirmed by alanine scanning (Ta[ble](#page-5-0) 2, entries 3−5). When the side chain of D-His at the AA^7 position was protected by a trityl group, the catalyst showed a lower reactivity (Table 2, entry 6). It is assumed that D-His of peptide 6 simply does not function as a bulky residue,

CH(CO₂Me)₂

Table 2. Evaluation of Peptide Sequences from Second Screenings

 $peptide -$

^aThe value in parentheses indicates the conversion with 10 mol % of peptide catalyst. ^bThe symbol in parentheses indicates the absolute configuration of the major enantiomer. "Found by the screening of library A-2nd. ^dPya indicates L-3-(1-pyrenyl)alanine. ^eFound by the screening of library B-2nd.

but the imidazole group is actively engaged in a catalytic cycle. The residues at sixth and eighth positions are considered to have a role for fixing the histidyl side chain in an appropriate three-dimensional space. By exchanging the eighth Trp of peptide 6 with more bulky $L-3-(1-pyrenyl)$ alanine (Pya), a comparable reactivity with an improved enantioselectivity was observed (Table 2, entry 7). Meanwhile, the screening of library B-2nd afforded the sequences highly enriched with D-Pro and D-Leu resi[dues \(se](#page-2-0)e Figure S2 in the Supporting Information). The detected sequences showed slightly higher conversions and ee values for the Michael reaction ([Table 2, entries 8 and 9\)](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.6b01591/suppl_file/jo6b01591_si_001.pdf), compared to peptide 5 (Table 1, entry 15). The enhancement of the reaction efficiency by the s[econd sc](#page-2-0)reening is not so remarkable, and it see[ms that s](#page-2-0)uch a long peptide chain is unnecessary. However, when the C-terminal residues of peptide 8 were truncated, the conversion was gradually decreased as the peptide shortened (Table 2, entries 10−12).¹²

Thus, two types of catalysts, peptides 7 and 8, with opposite enantioselectivities [were su](#page-2-0)ccessfully devel[ope](#page-5-0)d through the screenings and optimization. These peptides possess D- or Lhistidine at the seventh residue, and this amino acid is considered to be important for promoting the reaction. In our previous study, it was suggested that a histidine-containing prolyl peptide had the ability to bind an α , β -unsaturated aldehyde through the formation of an iminium intermediate at the N-terminal proline followed by the Michael addition of the imidazole group. $5,13$ In addition to the acceleration of the reaction by entrapping the substrate into resin beads, high enantioselectivity [ca](#page-5-0)n be expected because one face of the substrate is covered by the histidyl residue. To examine such abilities of the present catalysts, the substrate aldehyde and the resin-supported peptides were mixed, and the amount of the aldehyde remaining in solution phase was monitored by ${}^{1}H$ NMR (Figure 4). The entrapment of the aldehyde into the resin beads was observed, and a more reactive peptide in the Michael addition showed a faster uptake of the substrate.

Figure 4. Evaluation of substrate-binding capability of peptides.

Finally, other substrates were tested for the asymmetric Michael addition of a malonate (Table 3). Peptide 8 was chosen for the examination due to the highest enantioselectivity. In the presence of 2 mol % of the catalyst, the reaction proceeded sufficiently and afforded the product with excellent enantioselectivity (Table 3, entries 1−6).

In conclusion, efficient and enantioselective peptide catalysts for Michael addition of a malonate were developed by the screening of fully randomized libraries followed by the optimization with second-generation libraries. Two types of

Table 3. Asymmetric Michael Addition of a Malonate with Peptide 8

сно R (0.4 M) ÷ MeO OMe $(3$ equiv)		Pro-D-Pro-Aib-Tyr-D-Leu-D-Pro-His- -(D-Pro) ₃ -D-Leu-Leu-D-Leu- $8(2 \text{ mol } %)$ THF, 20 °C, 60 h						
entry	R	9	yield $(\%)$	ee $(\%)$				
$\mathbf{1}$	$4-NO_2C_6H_4$	a	84	99				
$\mathfrak{2}$	$4-CIC6H4$	b	83	99				
3	$4-BrC6H4$	c	80	98				
$4^{\mathfrak{a}}$	4 -MeOC ₆ H ₄	d	81	96				
5	C_6H_5	e	81	98				
6	2-naphthyl	\mathbf{f}	81	97				
a The concentration of the aldehyde was 0.6 M.								

peptides obtained in the screenings showed an opposite enantioselectivity to each other, although they possessed common N-terminal residues. A rational design for such sequences and optimization of them in a conventional way are considered to be difficult. This study demonstrates the potency of the present screening method, and further applications to other reactions are expected.

EXPERIMENTAL SECTION

Procedure for Constructing Peptide Libraries. For resin beads, TentaGel MB NH₂ (Rapp Polymere GmbH, product number: MB 250002, 0.23 mmol/g amine loading) was used. The coupling reaction of an amino acid was performed in an empty column with a filter at the bottom. Before the reaction, the resin was swollen with dichloromethane (DCM) for 20 min and washed with N,N-dimethylformamide (DMF). A typical procedure for the elongation of an amino acid was as follows. A DMF solution of an Fmoc-protected amino acid (3.0 equiv), O-benzotriazol-1-yl-N,N,N',N'-tetramethyluronium hexafluorophosphate (HBTU, 3.0 equiv), 1-hydroxybenzotriazole (HOBt, 3.0 equiv), and diisopropylethylamine (DIEA, 6.0 equiv) was added to the resin. The resulting mixture was stirred every 10 min with a vortex mixer. After 1 h, the resin was washed with DMF. To remove the Fmoc group, the resin was washed once with a DMF solution of piperidine (20 vol %), and this DMF/piperidine solution was added afresh to the resin. The resulting mixture was stirred in every 5 min with a vortex mixer. After 20 min, the resin was washed with DMF and used for the next coupling.

Preparation of Library A. After the coupling of methionine and the removal of its Fmoc group, Fmoc-NH– $(CH_2-CH_2-O)_8-$ (CH2)2−CO2H (Fmoc−OEG−OH, ChemPep, Inc., product number: 280112) was introduced. The coupling of this amino acid was performed with 2 equiv of Fmoc−OEG−OH, HBTU, and HOBt, and 4 equiv of DIEA for 4 h. After the removal of the Fmoc group, the resin beads were divided into six groups, and they were transferred into different columns. For these groups, Fmoc-protected Aib, His(Trt), Leu, D -Pro, Trp, and Tyr(tBu) were attached. After the coupling reactions, all beads were combined together, and the terminal Fmoc protection was removed. The series of the above manipulations, splitting the beads, coupling amino acids, mixing the beads, and removing the Fmoc group, was repeated until six randomized residues were introduced. Finally, Fmoc−Pro−OH was attached to the combined mixture of the beads. After the removal of the terminal Fmoc group, the resin was washed with DCM and dried under reduced pressure. To remove the protecting groups of side chains, the resin was soaked in trifluoroacetic acid containing water (3 vol %), and the resulting mixture was stirred every 10 min with a vortex mixer. After 1 h, the resin was successively washed with DCM, DMF, triethylamine/DCM (1:1), DMF, and DCM and was dried under reduced pressure.

Preparation of Library B. The same procedure for the preparation of library A was conducted using Aib, D-Leu, Leu-d₁₀, D-Pro, Trp, and $Tyr(tBu)$ as the constituent amino acids for the randomizing part.

Preparation of Library A-2nd. After the introduction of methionine and OEG and the removal of the Fmoc group according to the above procedure, the resin beads were divided into 10 groups, and they were coupled with Fmoc-protected Aib, Ala, Glu(OtBu), Gly, His(Trt), Leu, Pro, Thr(tBu), Trp, and Tyr(tBu). All beads were combined together, and the terminal Fmoc protection was removed. For the randomized positions of the library, amino acids were introduced in this way. For the seventh residue from the N-terminus, Aib, Gly, and the D-forms of Ala, Glu(OtBu), His(Trt), Leu, Pro, $Thr(tBu)$, Trp, and Tyr (tBu) were used. The common sequences, Pro-D-Pro at the first and second residues and D-Pro-Aib at the fourth and fifth residues, were introduced by the coupling reaction to the combined mixture of the beads. After the introduction of all residues, the following procedure was the same as above.

Preparation of Library B-2nd. Instead of N-terminal proline of library B, the common six residues were introduced to the randomized part using Fmoc-protected His(Trt), D-Pro, D-Leu, Tyr(tBu), Aib, D-Pro, and Pro, successively.

Procedure for Screening Peptide Libraries. To a roundbottom flask that contained a peptide library on resin beads (0.3 g, approximately 0.05−0.06 mmol of the terminal prolyl group), a solution of 4-nitrocinnamaldehyde and dye-labeled malonate 1 (10 mM for each reagent) in THF (5 mL) was added. After this mixture was stirred for 3 h in the initial screening, or for 10 min in the second screening, sodium triacetoxyborohydride (0.25 mmol) was added. After the resulting mixture was stirred for an additional 15 min, the resin beads were transferred to an empty column with a filter at the bottom and washed with DMF and DCM, successively. Then, the resin beads were put on a glass plate, and highly colored beads were picked up with a needle under a microscopic view. Each selected bead was put into an Eppendorf tube separately. To detach peptides from the beads, an aqueous solution (8 μ L) of formic acid (70%) and cyanogen bromide (50 mg/mL) was added to the tube, and this was left in the dark at 25 °C for 24 h. The solvent was removed under a stream of nitrogen gas and completely dried under reduced pressure. The residue was dissolved in methanol (100 μ L), and the solution was passed through a membrane filter (pore size: 0.45μ m). This sample was analyzed by an ESI-MS/MS measurement in a positive-ion mode.

Preparation of Resin-Supported Peptide Catalysts. A similar procedure with the preparation of peptide libraries was employed. For resin beads, TentaGel S NH₂ (AnaSpec, Inc., product number: 22798, 0.25 mmol/g amine loading) was used. Instead of HBTU and HOBt, O-(7-azabenzotriazol-1-yl)-N,N,N′,N′-tetramethyluronium hexafluorophosphate and 1-hydroxy-7-azabenzotriazole were used. The reaction time for the coupling was 30 min with stirring every 5 min, and the completion of the reaction was checked by the Kaiser test. The coupling reaction to an N-terminal Aib, L-, or D-Pro residue was performed twice without the Kaiser test.

Procedure for the Evaluation of Peptide Catalysts (Tables 1 and 2). To a mixture of a resin-supported peptide (0.006 or 0.003 mmol of the terminal prolyl group), 4-nitrocinnamaldehyde (0.03 mmol), and THF (300 μ L) was added dimethyl malon[ate \(0.09](#page-2-0) mmo[l\)](#page-2-0). After the mixture was stirred at 25 °C for 12 h, the peptide catalyst was filtered off and washed with chloroform. The conversion was determined by ¹H NMR measurement of the crude mixture in CDCl3. The ratio of the starting aldehyde and the product was used for the calculation. Then, the crude mixture was dissolved in t-butyl alcohol (200 μ L), 1 M aqueous solution of sodium dihydrogen phosphate (200 μ L), and 1 M aqueous solution of potassium permanganate (200 μ L). After the mixture was stirred for 30 min, 1 M aqueous solution of hydrochloric acid was added. The mixture was extracted with ethyl acetate, and the organic layer was washed with water three times. After the removal of the solvent under reduced pressure, the residue was dissolved in methanol (500 μ L) and toluene (200 μ L). To this solution was added trimethylsilyldiazomethane (0.6 M solution in hexane, 200 μ L). After the mixture was stirred for 10

min, the solvent was removed under reduced pressure. The residue was purified by preparative TLC (ethyl acetate/hexanes 1:1) to afford the corresponding methyl ester of the product.^{10b} Enantiomeric excess of this compound was determined by HPLC analysis with Chiralpak IA (hexane/2-propanol 7:3, 0.8 mL min[−]¹).5

Procedure for the Peptide-Catalyzed [Mic](#page-5-0)hael Addition of a Malonate (Table 3). To a mixture of resin[-s](#page-5-0)upported peptide 8 (8.7) mg, 0.0016 mmol of the terminal prolyl group), α , β -unsaturated aldehyde (0.08 mmol), and THF (200 μ L, or 133 μ L for 4methoxycin[namaldeh](#page-3-0)yde) was added dimethyl malonate (0.24 mmol). After the mixture was stirred at 20 $^{\circ}$ C for 60 h, the peptide catalyst was filtered off and washed with chloroform. The solvent was removed under reduced pressure, and the residue was purified by flash column chromatography on silica gel (ether/hexanes 3:7 to 9:1) to afford the Michael product. The spectral data of all products $({\bf 9a}_i^{\rm 10d}~{\bf 9b}_i^{\rm 10c}~{\bf 9c}$ e, 10b 9f⁵) were reported. To determine the enantioselectivity, Michael products were converted to the corresponding methyl e[ster](#page-5-0)s a[ccor](#page-5-0)ding t[o th](#page-5-0)e [pr](#page-5-0)evious report. $5,10$

■ ASSOCIATED [CON](#page-5-0)TENT

S Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.joc.6b01591.

Results of second-generation screenings, ESI-MS/MS [spectra of peptides,](http://pubs.acs.org) NMR s[pectra of Michael produc](http://pubs.acs.org/doi/abs/10.1021/acs.joc.6b01591)ts, and HPLC traces (PDF)

■ AUTHOR INFORM[ATIO](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.6b01591/suppl_file/jo6b01591_si_001.pdf)N

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Notes

The auth[ors declare no competin](mailto:kkudo@iis.u-tokyo.ac.jp)g financial interest.

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■ REFERENCES

(1) For reviews, see: (a) Davie, E. A. C.; Mennen, S. M.; Xu, Y.; Miller, S. J. Chem. Rev. 2007, 107, 5759. (b) Wennemers, H. Chem. Commun. 2011, 47, 12036.

(2) For recent selected examples, see: (a) Wiesner, M.; Upert, G.; Angelici, G.; Wennemers, H. J. Am. Chem. Soc. 2010, 132, 6. (b) Hrdina, R.; Müller, C. E.; Wende, R. C.; Wanka, L.; Schreiner, P. R. Chem. Commun. 2012, 48, 2498. (c) Kastl, R.; Wennemers, H. Angew. Chem., Int. Ed. 2013, 52, 7228. (d) Pathak, T. P.; Miller, S. J. J. Am. Chem. Soc. 2013, 135, 8415. (e) Zheng, L.; Marcozzi, A.; Gerasimov, J. Y.; Herrmann, A. Angew. Chem., Int. Ed. 2014, 53, 7599. (f) Yoganathan, S.; Miller, S. J. J. Med. Chem. 2015, 58, 2367. (g) Wende, R. C.; Seitz, A.; Niedek, D.; Schuler, S. M. M.; Hofmann, C.; Becker, J.; Schreiner, P. R. Angew. Chem., Int. Ed. 2016, 55, 2719. (3) (a) Copeland, G. T.; Miller, S. J. J. Am. Chem. Soc. 2001, 123, 6496. (b) Krattiger, P.; McCarthy, C.; Pfaltz, A.; Wennemers, H. Angew. Chem., Int. Ed. 2003, 42, 1722. (c) Krattiger, P.; Kovasy, R.; Revell, J. D.; Ivan, S.; Wennemers, H. Org. Lett. 2005, 7, 1101. (d) Matsumoto, M.; Lee, S. J.; Waters, M. L.; Gagné, M. R. J. Am. Chem. Soc. 2014, 136, 15817. (e) Matsumoto, M.; Lee, S. J.; Gagné, M. R.; Waters, M. L. Org. Biomol. Chem. 2014, 12, 8711.

(4) For other examples of library screenings, see: (a) Francis, M. B.; Jamison, T. F.; Jacobsen, E. N. Curr. Opin. Chem. Biol. 1998, 2, 422. (b) Berkessel, A. Curr. Opin. Chem. Biol. 2003, 7, 409. (c) Revell, J. D.; Wennemers, H. Curr. Opin. Chem. Biol. 2007, 11, 269. (d) Maillard, N.; Biswas, R.; Darbre, T.; Reymond, J.-L. ACS Comb. Sci. 2011, 13, 310. (e) Lichtor, P. A.; Miller, S. J. Nat. Chem. 2012, 4, 990.

(f) Sambasivan, R.; Ball, Z. T. Angew. Chem., Int. Ed. 2012, 51, 8568. (g) Giuliano, M. W.; Lin, C.-Y.; Romney, D. K.; Miller, S. J.; Anslyn, E. V. Adv. Synth. Catal. 2015, 357, 2301. (h) Gilbertson, S. R.; Collibee,

S. E.; Agarkov, A. J. Am. Chem. Soc. 2000, 122, 6522. (i) Tanaka, F.; Fuller, R.; Barbas, C. F., III Biochemistry 2005, 44, 7583.

(5) Akagawa, K.; Sakai, N.; Kudo, K. Angew. Chem., Int. Ed. 2015, 54, 1822.

(6) For reviews for aminocatalysis, see: (a) Erkkila, A.; Majander, I.; ̈ Pihko, P. M. Chem. Rev. 2007, 107, 5416. (b) Mukherjee, S.; Yang, J. W.; Hoffmann, S.; List, B. Chem. Rev. 2007, 107, 5471.

(7) (a) Furka, Á.; Sebestyén, F.; Asgedom, M.; Dibó, G. Int. J. Pept. Protein Res. 1991, 37, 487. (b) Lam, K. S.; Salmon, S. E.; Hersh, E. M.; Hruby, V. J.; Kazmierski, W. M.; Knapp, R. J. Nature 1991, 354, 82. (c) Lam, K. S.; Lebl, M.; Krchňák, V. Chem. Rev. 1997, 97, 411.

(8) The number of the beads classified as the most highly colored group was less than 100.

(9) To discriminate L- or D-leucine by mass spectrometry, deuterated leucine for the L-form was used.

(10) For examples of asymmetric Michael reactions of malonates to unsaturated aldehydes by aminocatalysts, see: (a) Yamaguchi, M.; Shiraishi, T.; Hirama, M. Angew. Chem., Int. Ed. Engl. 1993, 32, 1176. (b) Brandau, S.; Landa, A.; Franzén, J.; Marigo, M.; Jørgensen, K. A. Angew. Chem., Int. Ed. 2006, 45, 4305. (c) Maltsev, O. V.; Kucherenko, A. S.; Zlotin, S. G. Eur. J. Org. Chem. 2009, 2009, 5134. (d) Fleischer, I.; Pfaltz, A. Chem. - Eur. J. 2010, 16, 95. (e) Ghosh, S. K.; Dhungana, K.; Headley, A. D.; Ni, B. Org. Biomol. Chem. 2012, 10, 8322.

(11) A possible reason why the peptide with D-His at the seventh residue was not detected other than peptide 6 is that the ratio of highly active sequences in library A-2nd was much higher than that in the initial library. Changing screening conditions such as the concentration and the amount of the substrate might improve the screening sensitivity.

(12) Some other results with the sequences obtained from the second-generation libraries are shown in the Supporting Information.

(13) For an example of a related intermediate, see: Maltsev, O. V.; Chizhov, A. O.; Zlotin, S. G. Chem. - Eur. J. 2011, 17[, 6109.](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.6b01591/suppl_file/jo6b01591_si_001.pdf)